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Using the EMEP intensive
measurement periods to evaluate a
new parameterization of coarse
nitrate

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Outline

- Coarse nitrate formation
- Old parameterization
- New kinetic parameterization
 - Description of the method
 - Evaluation of model results
- New equilibrium parameterization
 - Description of the method
 - Evaluation of model results



Coarse nitrate formation on dust and sea-salt

- Coarse nitrate forms when nitric acid reacts with sea-salt and/or dust:



- The most reactive parts in dust are calcite ($\text{Ca}(\text{CO}_3)$) and dolomite ($\text{CaMg}(\text{CO}_3)_2$)



Coarse nitrate in the model today

- Coarse nitrate in the EMEP model only depends on HNO_3 , with a reaction rate coefficient k_{RH}

$$k_{\text{RH}} = 1.0 \times 10^{-4} \text{ for RH} > 90\%$$

$$k_{\text{RH}} = 5.0 \times 10^{-6} \text{ for RH} < 90\%$$

- The model run with this parameterization is later referred to as E_std



New kinetic parameterization

- Kinetic method: assumed total reaction in forward direction of reactions 1 and 2, with a reaction rate that depends on aerosol surface and the sticking coefficient γ of HNO_3

$$k = \left(\frac{d}{2D_g} + \frac{4}{\nu\gamma} \right)^{-1} A$$



The different model runs

- E_std: model run with old parameterization
- E_d: model run with reactions on dust only
 - E_d_ca60
- E_ss: model run with reactions on sea-salt only
- E_d_ss: model run with reactions on both dust and sea-salt

Coarse nitrate in model results compared to measurements



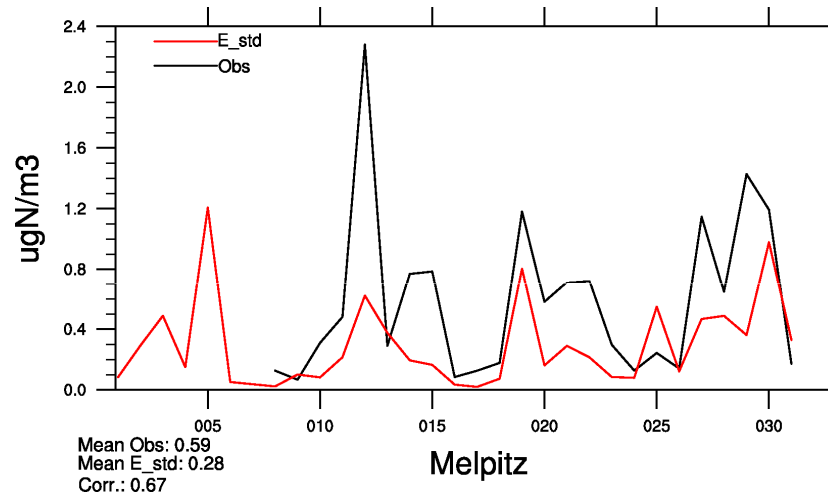
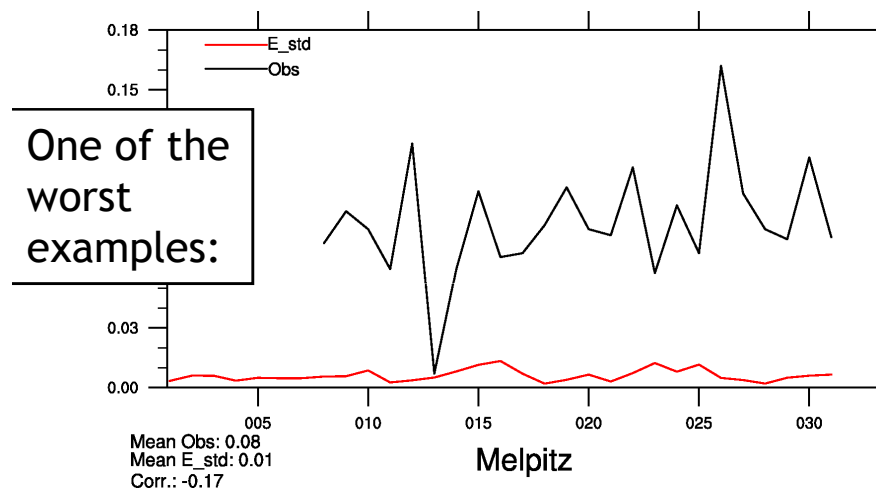
Station ID	Obs $\mu\text{g(N)} \text{ m}^{-3}$	Bias (%)				Correlation				
		E_std	E_ss	E_d	E_d_ss	E_std	E_ss	E_d	E_d_ss	
June 2006										
CH02	0.10	-60	-100	-97	-96	-0.28	-0.03	0.56	0.53	
GB36*	0.51	-71	-96	-100	-94	0.01	0.39	-0.18	0.39	
NL11*	0.03	67	-63	-98	-61	0.22	-0.15	0.32	-0.12	
FI17	0.10	-3	-92	-99	-92	0.35	0.66	0.00	0.66	
IT01	0.36	-64	-97	-94	-91	0.17	-0.06	0.36	0.22	
NO01	0.11	-54	-91	-100	-91	0.64	0.62	0.13	0.62	
DE44	0.12	-33	-96	-99	-96	0.17	0.56	-0.13	0.54	
ES17	0.20	-55	-95	-85	-79	0.16	-0.46	0.84	0.75	
January 2007										
CH02	0.29	-79	-97	-100	-98	0.39	0.52	0.30	0.53	
NL11*	0.08	-64	-89	-100	-89	0.11	0.06	0.13	0.06	
FI17	0.06	-27	-96	-100	-95	-0.03	0.05	0.07	0.05	
IT01	0.30	-30	-87	-99	-85	0.06	0.22	0.00	0.21	
NO01	0.002	-21	-74	-95	-74	0.22	0.33	0.14	0.33	
DE44	0.10	-50	-90	-100	-95	0.24	0.39	0.15	0.39	

- Temporal correlation improves
- There is a negative bias in the new parameterization



5% Ca^{2+} in coarse dust compared to measured (PM10-PM1) Ca^{2+} , 2007

Na^+ in coarse sea-salt compared to measured (PM10-PM25) Na^+ , 2007

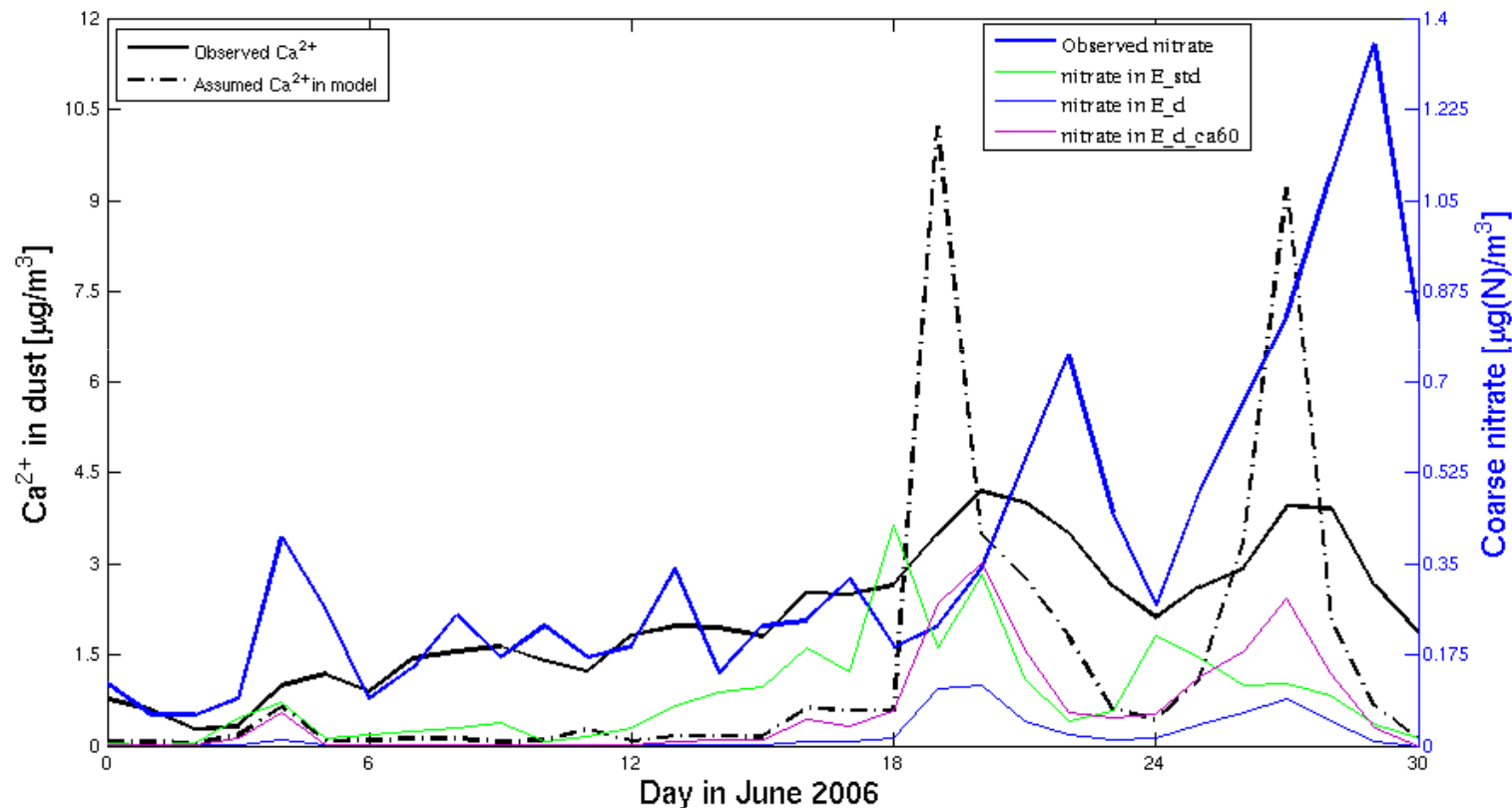


- Can the low value of 5% Ca^{2+} in coarse dust compared with measurements explain the high negative bias?
- Underestimation of coarse sea-salt may be a reason for underestimating coarse nitrate formation



Coarse nitrate on dust, 2006

Montelibretti (IT01)



- Assuming more Ca^{2+} content in dust did not explain the negative bias in the dust parameterization

Equilibrium method



- Assume that formation of coarse nitrate on sea-salt and dust reaches equilibrium within the 20-minute time step in the EMEP model
- Use the equilibrium module EQSAM to calculate the equilibrium
- First let the photochemistry work, then let fine aerosols reach equilibrium, before letting coarse aerosols reach equilibrium
- This model run is later referred to as E_eqsam



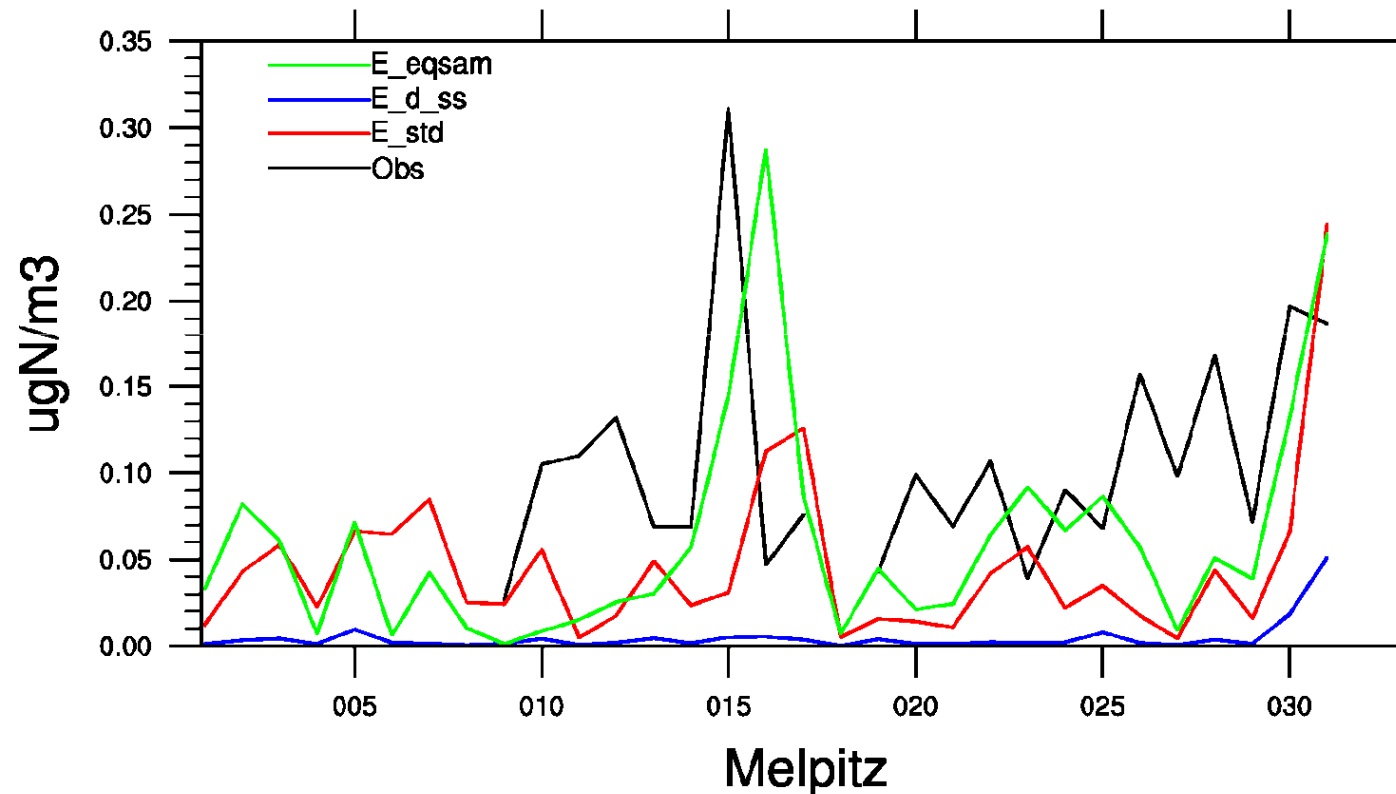
Station ID	Obs $\mu\text{g(N)} \text{ m}^{-3}$	Bias (%)			Correlation		
		E_std	E_d_ss	E_eqsam	E_std	E_d_ss	E_eqsam
June 2006							
CH02	0.10	-60	-96	-90	-0.28	0.53	-0.20
GB36*	0.51	-71	-96	-82	0.01	0.39	-0.02
NL11*	0.03	67	-67	89	0.22	-0.12	-0.04
FI17	0.10	-3	-88	-96	0.35	0.77	0.05
IT01	0.36	-64	-91	-100	0.17	0.22	-0.20
NO01	0.11	-54	-91	-91	0.64	0.62	0.20
DE44	0.12	-33	-92	-92	0.17	0.54	0.41
ES17	0.20	-55	-80	-95	0.16	0.75	0.60
January 2007							
CH02	0.29	-79	-98	-76	0.39	0.53	0.56
NL11*	0.08	-64	-89	-12	0.11	0.06	0.30
FI17	0.06	-27	-95	-27	-0.03	0.05	0.00
IT01	0.30	-30	-85	-27	0.06	0.21	-0.05
NO01	0.002	-21	-74	-37	0.22	0.33	0.41
DE44	0.10	-50	-95	-30	0.24	0.39	0.33

- The equilibrium method improves temporal correlation and the bias in winter



Coarse nitrate at Melpitz in 2007

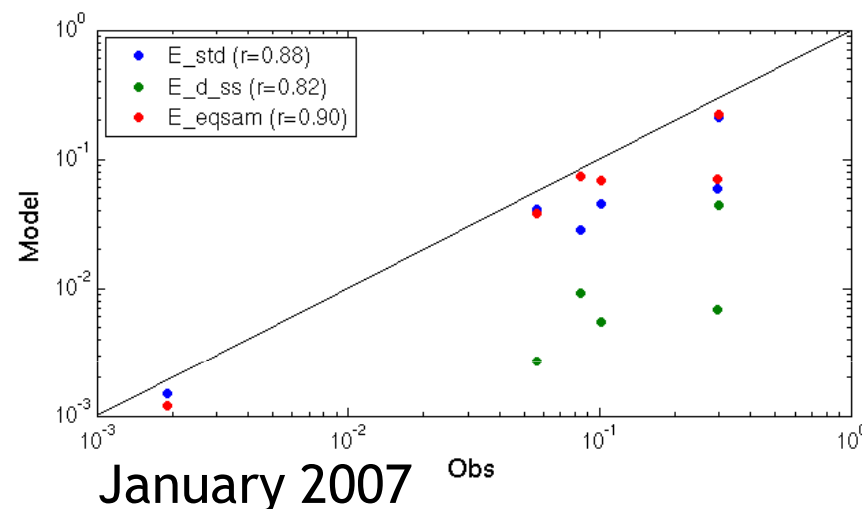
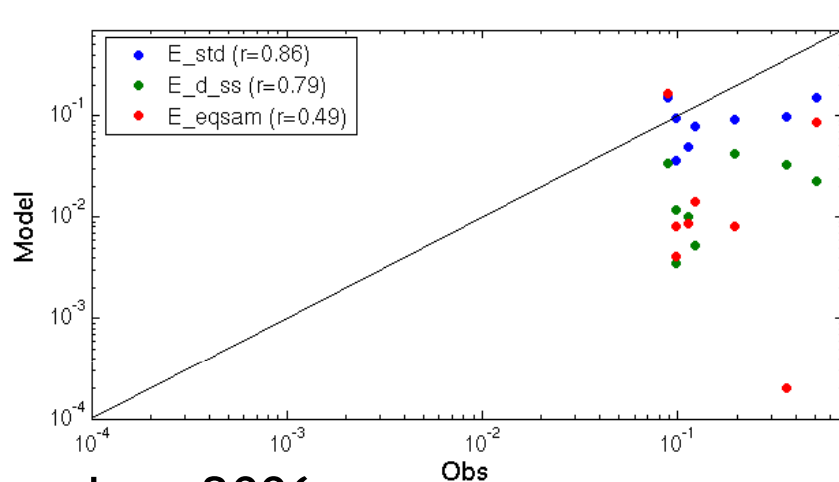
pm10-pm25 Nitrate in air



- The equilibrium method improves both correlation and bias at Melpitz (DE44)



Spatial correlation of coarse nitrate



- Good spatial correlation, especially in January
- Small improvement in spatial correlation indicates that the formation of coarse nitrate is mainly driven by HNO_3
- The spatial correlation of dust and sea-salt limits the spatial correlation of coarse nitrate



Summary

- Inclusion of coarse nitrate formation on dust and sea-salt with the kinetic approach improves the temporal correlation but gives a high underestimation of coarse nitrate
 - Several sensitivity tests were performed, but neither of them could explain this high negative bias
 - Comparison with other model studies of coarse nitrate formation on dust and sea-alt reveals a rather small production of coarse nitrate in this study
- Inclusion of coarse nitrate formation on dust and sea-salt assuming equilibrium improves the result in winter